

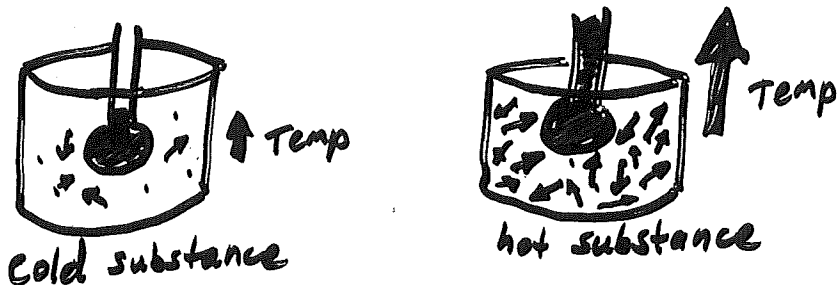
# Temperature

①

\* Temp. is defined as the measure of heat or energy that a substance has.

- The more energy a substance's molecules have - the higher the temp.

The thermometers that we use in the lab measure the temperature of substances using this principle.



When measuring temp. there are 3 main units that can be used.  $^{\circ}\text{F}$ ,  $^{\circ}\text{C}$  &  $\text{K}$  (Kelvin).

\* Each temp. system has a strong pt. based on ease of use w/ freezing pt. or other phase change pts. of  $\text{H}_2\text{O}$ .

ex)  $^{\circ}\text{C}$  has the freezing pt. of  $\text{H}_2\text{O}$  as  $0^{\circ}\text{C}$   
\* easy to remember

$\text{K}$  has absolute zero as  $0 \text{ K}$  (coldest theoretical temperature)

$^{\circ}\text{F}$  . . . . water freezes @  $32^{\circ}\text{F}$ , boils @  $212^{\circ}\text{F}$   
absolute zero =  $-460^{\circ}\text{F}$

\* Just like all measurements, temp. is used in different units depending on situation or culture. (2)

• If you are given a temp. in any system, you can convert to any other system using formulas below.

If given °C →  $^{\circ}\text{F} = \left(\frac{9}{5}^{\circ}\text{C}\right) + 32$

→  $\text{K} = ^{\circ}\text{C} + 273.15$

If given °F →  $^{\circ}\text{C} = (\text{F} - 32) \div \frac{9}{5}$

→  $\text{K} = \left[(\text{F} - 32) \div \frac{9}{5}\right] + 273.15$

If given K →  $^{\circ}\text{C} = \text{K} - 273.15$

→  $\left[(\text{K} - 273.15) \times \frac{9}{5}\right] + 32 = ^{\circ}\text{F}$

ex)  $29.5^{\circ}\text{C} \rightarrow ^{\circ}\text{F}$

ex)  $1,250\text{K} \rightarrow ^{\circ}\text{C}$

\* New Scale - Rankine =  $^{\circ}\text{F} + 460$

Not used much, but is the "zero" version of °F

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You need to apply the formulas to solve a 3x3 grid problem like the one below (give °F, °C, or K - solve for the rest)

°F	°C	K
		320 K
	82 °C	
65 °F		

\* Solve horizontally  
↔

Another

°F	°C	K
123 °F		
	180 °C	
		412 K